## 7-O-METHYLSPINOCHROME B AND ITS 6-(3-HYDROXY-*N*-BUTYL)-DERIVATIVE, CATECHOL-O-METHYL TRANSFERASE INHIBITORS, PRODUCED BY *FUNGI IMPERFECTI*

Sir:

In a previous paper<sup>1)</sup> we reported isolation of two catechol-O-methyl transferase (COMT) inhibitors produced by a *Streptmyces* and identified as methylspinazarin (2, 3, 5, 8-tetrahydroxy-6methyl-1, 4-naphthoquinone) and 6, 7-dihydromethylspinazarin. In this paper we report the isolation of other inhibitors which are produced by *Fungi imperfecti*. Their structures were determined to be 7-O-methylspinochrome B (2, 3, 5trihydroxy-7-methoxy-1, 4-naphthoquinone) and 6-(3-hydroxy-*n*-butyl)-7-O-methylspinochrome B, a new compound.

In the course of our screening for COMT inhibitors, cultured broth of a *Fungi imperfecti* obtained from Sanraku-Ocean Co., Ltd., Tokyo, showed strong inhibition against COMT by a modification of the assay method described by AXELROD *et al.*<sup>2)</sup> This strain was classified by Mr. K. KOHNO of Research Institute of Sanraku-Ocean Co., Ltd., to be *Corynespora cassi cola*. Maximum production was observed after 96 hours of the submerged culture in a medium containing 1.0% potato starch, 0.5% glucose, 1.0% soybean meal, 0.25% yeast extract, 0.05% KH<sub>2</sub>PO<sub>4</sub> and 0.025% MgSO<sub>4</sub>·7H<sub>2</sub>O, at 27°C.

The cultured broth was filtered, and the filtrate was extracted with butyl acetate at pH 2.0. The separated cake was extracted with methanol. The methanol extract was concentrated and extracted with butyl acetate at pH 2.0. The combined butyl acetate extract was dried under reduced pressure, yielding a reddish gummy material. After washed with petroleum ether, the residue was dissolved in acetone. The acetone-soluble fraction was dried to a reddish powder. The active agent was further purified by silica gel column chromatography with benzene-acetone (10:1). The activity appeared in two fractions. The most rapidly eluted compound (I), was further purified by Sephadex LH-20 column chromaotgraphy with methanol and then crystallized from acetonebenzene to give reddish needles, m. p. 267°C. The second compound (II) was isolated by the same procedure to give reddish orange needles,

m. p. 202°C.

Compound I has a molecular formula  $C_{11}H_8O_8$ (MW 236). Found: C, 56.34; H, 3.84; O, 39.82. Calcd.: C, 55.94; H, 3.41; O, 40.65.  $\nu_{\text{KBr}}$ : 3480 cm<sup>-1</sup> (OH), 1660, 1620 cm<sup>-1</sup> (perihydroxy quinone.  $\lambda_{\max}^{\text{EtOH}}(\log \varepsilon)$ : 272 nm (4.33), 285 (shoulder), 322 (3.99), 385 (3.56) and 485 (3.26). It gives positive ferric chloride (violet) and magnesium acetate (violet) tests for perihydroxy quinone. The NMR spectrum in DMSO-d<sub>6</sub> solution showed a methoxy signal at  $\delta$  3.88 and meta-coupled aromatic proton signals at  $\delta$  6.50 and 6.95 (J=2.5 Hz). Broad OH proton signals were centered at  $ca. \delta 10.8$  and a hydrogen bonded OH signal was seen at  $\delta$  12.10. Nuclear OVERHAUSER effects were observed between the methoxy and both aromatic proton signals, +12.0% CH<sub> $\delta 6.50$ </sub>{OCH<sub>3</sub>}, +12.0% CH<sub> $\delta 6.95$ </sub>  $\{OCH_3\}.$ 

Compound I gave a dimethyl ether by treatment with diazomethane, m.p. 113°C, M<sup>+</sup> m/e264 (C<sub>18</sub>H<sub>12</sub>O<sub>6</sub>). The NMR spectrum in CDCl<sub>3</sub> shows three methoxy signals at  $\delta$  3.87, 4.07 and 4.11, meta-coupled aromatic proton signals at  $\delta$  6.61 and 7.17 (J=2.5 Hz) and a hydrogen bonded OH signal at  $\delta$  12.08.

From the above data, I was suggested to be 7-O-methylspinochrome B (2, 3, 5-trihydroxy-7-methoxy-1, 4-naphthoquinone) and its dimethyl ether to be the 2, 3-dimethyl ether of I.

Both compounds are known compounds derived from spinochrome B, a pigment isolated from the sea urchin. Spinochrome B trimethyl ether<sup>3~6</sup> and its partially demethylated product<sup>6</sup> were identical with the dimethyl ether of I and compound I respectively, when compared with the melting point, UV and NMR spectra described in the literature<sup>3~6</sup>.

Compound II has a molecular formula of  $C_{15}H_{16}O_7$  (MW 308). Found: C, 58.14; H, 5.36; O, 36.50. Calcd.: C, 58.44; H, 5.23; O, 36.33.  $\nu_{\rm KBr}$ : 3400 cm<sup>-1</sup> (OH), 1650, 1610 cm<sup>-1</sup> (peri hydroxy quinone).  $\lambda_{\rm max}^{\rm EvOH}$  (log  $\varepsilon$ ): 273 nm (4.26), 286 (shoulder), 331 (3.87) and 400 (3.44). It gives the same color reactions as I.

By treatment with diazomethane, II gave a dimethyl ether, m. p.  $108 \sim 109^{\circ}$ C which had a hydrogen bonded OH ( $\delta$  12.38, singlet, in CDCl<sub>3</sub>). These data suggested that II and I have the same chromophore. The NMR studies of II indicated the presence of a 3-hydroxy-*n*-butyl substituent:  $\delta$  1.10 (3H, doublet, J=6 Hz),

3.50 (1H, multiplet), 1.50 (2H, multiplet) and 2.65 (2H, triplet, J=6 Hz) in DMSO-d<sub>6</sub>. There were also one methoxy ( $\delta$  3.85), one aromatic proton (7.06, singlet), broad OH (*ca.* 11.0) and one hydrogen bonded OH (12.65, singlet) peaks. A nuclear OVERHAUSER effect was observed between the methoxy and aromatic proton signals, +18.0% CH{OCH<sub>8</sub>}. The above results indicated that II is a 3-hydroxy-*n*-butyl derivative of I. There are two possibilities for the position of the alkyl side chain, C<sub>8</sub> or C<sub>8</sub>. The chemical shift of the aromatic proton ( $\delta$  7.06) indicated that the proton should be present at  $C_8$ . Thus it showed the presence of an alkyl substituent at  $C_6$ , because the chemical shift of an aromatic proton at  $C_6$  should be at around  $\delta 6.5^{4}$ ). Thus, we propose 2, 3, 5-trihydroxy-6-(3-hydroxy-n-butyl)-7-methoxy-1,4-naphthoquinone as the structure of II.

Compounds I and II inhibited COMT.  $ID_{50}$  values of compounds I and II were  $10 \mu g/ml$  (4.23×10<sup>-5</sup>M) and 7.0  $\mu g/ml$  (2.27×10<sup>-5</sup>M) respectively. LINEWEAVER-BURK plots of the kinetic data on both compounds I and II showed

Fig. 1. LINEWEAVER-BURK plot of adrenaline concentration against rate of O-methylation with or without compounds I and II

Reaction mixture contained 50  $\mu$  moles of phosphate buffer, pH 8.0, various concentrations of L-adrenaline, 2.5  $\mu$  moles of magnesium chloride, 0.375  $\mu$  moles of S-adenosyl-L-methionine-methyl-H<sup>3</sup> (1  $\mu$  Ci/ $\mu$  mole), enzyme preparation, and distilled water to final volume of 0.5 ml

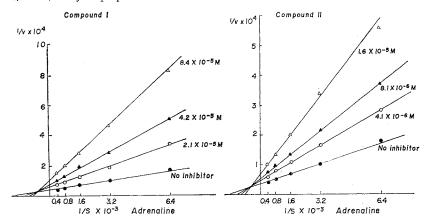
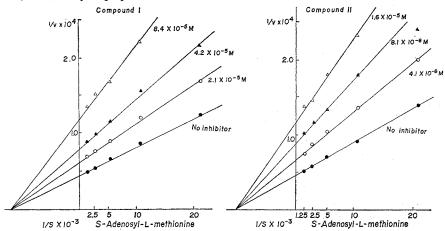


Fig. 2. LINEWEAVER-BURK plot of S-adenosyl-L-methionine concentration against rate of O-methylation with or without compounds I and II

Reaction mixture contained 50  $\mu$  moles of phosphate buffer, pH 8.0, 1.25  $\mu$  moles of L-adrenaline, 2.5  $\mu$  moles of magnesium chloride, various concentrations of S-adenosyl-L-methionine-methyl-H<sup>3</sup> (1  $\mu$  Ci/ $\mu$  mole), enzyme preparation, and distilled water to final volume of 0.5 ml



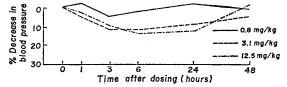
the mixed type of inhibition against adrenaline and the non-competitive type of inhibition against S-adenosyl-L-methionine (Figs. 1 and 2). The inhibition constants, Ki, values of compounds I and II against adrenaline were  $2.1 \times$  $10^{-5}$ M and  $6.2 \times 10^{-6}$ M respectively, while the MICHAELIS-MENTEN constant, Km, value of adrenaline was 6.3×10<sup>-4</sup>M. Against S-adenosyl-L-methionine, the Ki values of compounds I and II were  $4.2 \times 10^{-5}$  M and  $8.4 \times 10^{-6}$  M respectively, while the Km value of S-adenosyl-Lmethionine was  $8.1 \times 10^{-5}$  M. As inhibition by compounds I and II did not show any change at excess concentration of magnesium ion (2 $\times$  $10^{-2}$ M), the effect of these compounds is not due to chelation of magnesium ion.

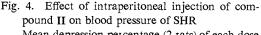
The effect of compounds I and II on tyrosine hydroxylase<sup>7)</sup> (TH) and dopamine- $\beta$ -hydroxylase<sup>8)</sup> (D $\beta$ H) was also tested. TH was inhibited by 100  $\mu$ g/ml (4.23×10<sup>-4</sup>M: ID<sub>50</sub>) of compound I and by 150  $\mu$ g/ml (4.87×10<sup>-4</sup>M; ID<sub>50</sub>) of compound II. These compounds did not inhibit D $\beta$ H at 100  $\mu$ g/ml.

The hypotensive effect was studied in the spontaneously hypertensive rat (SHR). The results observed with the intraperitoneal injection of 12.5 mg/kg, 3.1 mg/kg and 0.8 mg/kg are shown in Figs. 3 and 4. From these results, it is clear that compound II markedly decreases

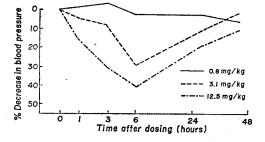
Fig. 3. Effect of intraperitoneal injection of compound I on blood pressure of SHR

Mean depression percentage (2 rats) of each dose





Mean depression percentage (2 rats) of each dose



blood pressure in the SHR.

No toxicity was observed after the intraperitoneal injection of 200 mg/kg of each compound to mice. These compounds at  $100 \mu \text{g/ml}$  showed no antibacterial activity.

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